

**Republic of Yemen**  
**Ministry of Higher Education & Scientific Research**  
**Emirates International University**



**Faculty of Medicine & Health Sciences**  
**Department of pharmacy**

**Bachelor of Pharm D**

**Course Specification of**  
**Physical Pharmacy**

**Course No.(PHYP 105)**



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Prepared by:

~~Dr. Sami Ahmed Alsamir~~

Reviewed by:

Dr. Mokhtar  
Almorani

Head of the Department:

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Quality Assurance Head

Dean:



## I. Course Identification and General Information:

1	Course Title:	Physical Pharmacy			
2	Course Code & Number:	PHYP 105			
3	Credit Hours:	Credit Hours	TheoryHours		Lab. Hours
			Lecture	Exercise	
		2	2	--	--
4	Study Level/ Semester at which this Course is offered:	1 <sup>st</sup> Level / 2 <sup>nd</sup> Semester			
5	Pre –Requisite (if any):	-			
6	Co –Requisite (if any):	None			
7	Program (s) in which the Course is Offered:	Bachelor of Pharm D			
8	Language of Teaching the Course:	English			
9	Study System:	Semester based System			
10	Mode of Delivery:	Full Time			
11	Location of Teaching the Course:	Faculty of Medicine & Health Sciences			
12	Prepared by:	Dr/ Sami Ahmed Al Sammit			
13	Date of Approval:				

## II. Course Description:

The course aims to provide students with basic principles of the physicochemical properties of pharmaceutical and various substances that are used in designing and preparation of different pharmaceutical dosage forms.

<b>III. Course Intended Learning Outcomes (CILOs) : (maximum 8)</b> <b>Upon successful completion of the course, students will be able to:</b>		<b>Referenced PILOs</b> <b>Learning out of program</b>		
<b>A. Knowledge and Understanding:</b>		<b>I,A</b> <b>or E</b>		
a1	Illustrate the physicochemical properties of pharmaceutical and various substances which are used in preparation of different pharmaceutical dosage forms.	I		A1
a2	Outline properties of different pharmaceutical dosage forms and drug delivery systems.	I		A3
a3	Discuss the principles of adsorption/desorption, interfacial tensions, and emulsifying, suspending and surface active agents.	E		A11
a4	Identify principles and types of rheology.	E		
a5	Explain the reaction kinetics and drug stability	E		
<b>B. Intellectual Skills:</b>				
b1-	Relate the effect of physicochemical properties on formulation of pharmaceutical dosage forms.	E		B1
b2-	Predict the order of reaction kinetics and drug degradation pathways	E		B3
b3-	Explore the basics fundamentals of physical pharmacy in designing of different pharmaceutical dosage forms.	I		B1
<b>C. Professional and Practical Skills:</b>				



c1-	Apply the rheological properties of some pharmaceutical substance to develop pharmaceutical preparations	A		C1
c2-	Practice the solubility, partition coefficient and surface tension of some pharmaceutical substance.	A		C6
c3-	Test the proper storage conditions based on drug degradation pathway	A		C6
c4-	Calculate the reaction kinetic order and shelf life of some pharmaceutical substances	A		C6
<b>D. Transferable Skills:</b>				
d1-	Implement writing and presentation skills and demonstrate creativity and time management.	A		D2
d2	Work independently or collaboratively as a teamwork member to prepare seminars/ presentations or write reports	A		D3

**(A) Alignment of Course Intended Learning Outcomes (Knowledge and Understanding) to Teaching Strategies and Assessment Methods:**

Course Intended Learning Outcomes	Teaching Strategies	Assessment Strategies
a1 Illustrate the physicochemical properties of pharmaceutical and various substances which are used in preparation of different pharmaceutical dosage forms.	▪ Lectures	Final Written Exam, homework, report,
a2 Outline properties of different pharmaceutical dosage forms and drug delivery systems.	▪ Lectures.	▪ Final Written Exam, homework, report, Quizzes
a3 Discuss the principles of adsorption/ desorption, interfacial tensions, and emulsifying, suspending and surface active agents.	▪ Lectures, Problem-based learning.	▪ Exam, homework, report, Quizzes
a4 Identify principles and types of	▪ Lectures, Discussions, Problem-based learning.	▪ Final Written Exam,

	rheology.	based learning.	homework, report,
a5	Explain the reaction kinetics and drug stability	▪ Lectures , Small group discussions.	▪ Exam, homework, Quizzes.
<b>(B) Alignment of Course Intended Learning Outcomes (Intellectual Skills) to Teaching Strategies and Assessment Methods:</b>			
	<b>Course Intended Learning Outcomes</b>	<b>Teaching Strategies</b>	<b>Assessment Strategies</b>
b1	Relate the effect of physicochemical properties on formulation of pharmaceutical dosage forms.	Lectures, Discussions, brain storming, Solving Problem methods	▪ Exam, Quizzes,
b2	Predict the order of reaction kinetics and drug degradation pathways	▪ Lectures, Discussions, brain storming	▪ Homework, short answers and Written exam
b3	Explore the basics fundamentals of physical pharmacy in designing of different pharmaceutical dosage forms.	▪ Lectures, Discussions.	▪ Exam, Quizzes, Homework
<b>(C) Alignment of Course Intended Learning Outcomes (Professional and Practical Skills) to Teaching Strategies and Assessment Methods:</b>			
	<b>Course Intended Learning Outcomes</b>	<b>Teaching Strategies</b>	<b>Assessment Strategies</b>
c1-	Apply the rheological properties of some pharmaceutical substance to develop pharmaceutical preparations	▪ Lectures and Group discussion	▪ Homework, Written exam
c2-	Practice the solubility, partition coefficient and surface tension of some pharmaceutical substance.	▪ Group learning and Problem-based learning.	Homework, Written exam
c3-	Test the proper storage conditions based on drug degradation pathway	▪ Lectures and brain storming	Homework, Written exam
c4-	Calculate the reaction kinetic order and shelf life of some pharmaceutical substances	▪ Lectures and brain storming	▪ Homework, Written exam
<b>(D) Alignment of Course Intended Learning Outcomes (Transferable Skills) to Teaching Strategies and Assessment Methods:</b>			
	<b>Course Intended Learning Outcomes</b>	<b>Teaching Strategies</b>	<b>Assessment Strategies</b>
d1	Implement writing and presentation	▪ Lectures	▪ Homework



	skills and demonstrate creativity and time management.	<ul style="list-style-type: none"> <li>Group Working</li> </ul>	
d2	Work independently or collaboratively as a teamwork member to prepare seminars/presentations or write reports.	<ul style="list-style-type: none"> <li>Group Working</li> <li>Problem Solving</li> </ul>	<ul style="list-style-type: none"> <li>Homework</li> </ul>

#### IV. Course Contents:

##### A. Theoretical Aspect:

No.	Units/Topics List	Sub Topics List	Number of Weeks	Contact Hours	Learning Outcomes (CILOs)
1	Introduction to physical pharmacy	Definition, orientation, process	1	2	a1, a2, b1,b3, c2, d2
2	State of matter, Solids	Crystal structure and external appearance, polymorphism, crystal hydrates, wetting of solid surfaces and powders dissolution of drugs Solid dispersions	1	2	a1, a2, b1,b3, c1, d1,d2
3	Solubility and solution	Solvents for pharmaceutical aerosols, pH of drug solutions, Buffers	1	2	a1,a2,a3, b1,b3, c2, d1,d2
4	Factors influencing solubility	isotonic solutions, Diffusion of drugs in solution	1	2	a1,a2,a3, b1,b3, c2, d1,d2
5	Drug stability	Definition, factors stability of liquid and solids dosage forms	1	2	a1, a2, a5, b1, b2,b3, c3,c4, d1,d2
6	Reaction Kinetics and drug stability	Kinetics of chemical decomposition in solution Stability testing calculation of shelf-life	1	2	a1, a2, a5, b1, b2,b3, c3,c4, d1,d2

No.	Units/Topics List	Sub Topics List	Number of Weeks	Contact Hours	Learning Outcomes (CILOs)
7	Surface and interfacial tensions	Definition , factors affecting Surface and interfacial tensions	1	2	a1, a1, a3, b1, b3, c2, d1,d2
8	Mid-term exam		1	2	a1, a1, a3, a5, b1-3, c2-4
9	Surface active agents	Definitions Some typical surfactants, applications	1	2	a1, a1, a3, b1, b3, c2, d1,d2
10	Emulsions, suspensions and other dispersed systems	Foams and defoamers	1	2	a1, a1, a3, b1, b3, c2, d1,d2
11	Polymers, drug absorption	Properties, Solution properties of polymers, Routes of administration	1	2	a1,a2,a3, b1,b3, c2, d1,d2
12	Physicochemical drug interactions and incompatibilities Complexes; classification and use.	Solubility problems pH effects in vitro and in vivo, Analysis of complexes.	1	2	a1, a2, a3, a4, b3, c1-2, d1,d2
13	Peptides, proteins and other biopharmaceuticals	Structure and solution properties of peptides and proteins, The stability of proteins and peptides	1	2	a1,a2,a3, b1,b3, c2, d1,d2
14	Adsorption at solid and liquid interface.	Definition, types factors affecting Adsorption of drugs	1	2	a1, a1, a3, b1, b3, c2, d1,d2
15	Rheology	Definition , classification and Application of polymers in drug delivery, Rheological	1		a1, a2, a3, a5, b1, b3,



No.	Units/Topics List	Sub Topics List	Number of Weeks	Contact Hours	Learning Outcomes (CILOs)
		characteristics of products			c1, d2
16	Final Exam		1	2	a1-5, b1-3, c1-4
Number of Weeks /and Units Per Semester			16	32	

### V. Teaching Strategies of the Course:

Lectures , Discussions, Small group discussions, brainstorming, Group learning and Problem-based learning.

### VI. Assessment Methods of the Course:

Quizzes, Homework , short answers and Written exam

### VII. Assignments:

No.	Assignments	Week Due	Mark	Aligned CILOs (symbols)
1	Assignment 1: Home works	9 <sup>th</sup>	5	a2,a3, a5, b2,b3, c3, d2
2	Assignment 1: Reports	13 <sup>th</sup>	5	a1-4, b1 c3,c4,d1, d2
Total			10	

### VIII. Schedule of Assessment Tasks for Students During the Semester:



No.	Assessment Method	Week Due	Mark	Proportion of Final Assessment	Aligned Course Learning Outcomes
1	Assignments	9 <sup>th</sup> , 13 <sup>th</sup>	10	10%	a1-5, b1-3, c3,c4, d1,d2
2	Quizzes 1 & 2	2 <sup>th</sup> and 11 <sup>th</sup>	10	10%	a2, a3, a5, b1, b3, c2
3	Mid-Term Theoretical Exam	8 <sup>th</sup>	20	20%	a1, a1, a3, a5, b1-3, c2-4
4	Final Theoretical Exam	15 <sup>th</sup>	60	60%	a1-5, b1-3, c1-4
<b>Total</b>			<b>100</b>	<b>100%</b>	

<b>IX. Learning Resources:</b>	
<ul style="list-style-type: none"> <li>Written in the following order: Author, Year of publication, Title, Edition, Place of publication, Publisher.</li> </ul>	
<b>1- Required Textbook(s) ( maximum two ):</b>	
<ol style="list-style-type: none"> <li>Notes on Physical Pharmacy prepared by the department staff.</li> <li>Florence, A.T. and Attwood, D., 2008, "FASTtrack physical pharmacy" 1st edition, Pharmaceutical Press, London.</li> <li>Martin, A., 2006, 'Physical Pharmacy – physical chemical principles in pharmaceutical sciences' 5th edition, Lippincott Williams &amp; Wilkins., Philadelphia.</li> </ol>	
<b>1- Recommended Readings and Reference Materials</b>	
<ol style="list-style-type: none"> <li>Florence, A.T. and Attwood, D., 2006, "Physicochemical principles of pharmacy", 4<sup>th</sup> edition, Pharmaceutical Press, London.</li> <li>Loyd, V Allen J, 2013, Remington: The Science and Practice of Pharmacy 22<sup>nd</sup> edition, Pharmaceutical Press, London.</li> <li>Ansel; H.C, (2011) Pharmaceutical Dosage Forms and drug Delivery Systems'. 9<sup>th</sup>ed , Lea &amp;Febiger; Philadelphia; London.</li> <li>Aulton, M.E, (2013) Pharmaceutics, the design and manufacture of medicines. 4<sup>th</sup> edition, Churchill Livingstone, Edinburgh.</li> </ol>	
<b>3- Electronic Materials and Web Sites etc.:</b>	
<a href="http://www.pubmed.com">www.pubmed.com</a>  <a href="http://www.sciencedirect.com">http://www.sciencedirect.com</a>	

<b>X. Course Policies: (Based on the Uniform Students' By law (2007))</b>	
1	<b>Class Attendance:</b> Class Attendance is mandatory. A student is considered absent and shall be banned from taking the final exam if his/her absence exceeds 25% of total classes.
2	<b>Tardiness:</b> A student will be considered late if he/she is not in class after 10 minutes of the start time of class.
3	<b>Exam Attendance/Punctuality:</b> No student shall be allowed to the exam hall after 30 minutes of the start time, and shall not leave the hall before half of the exam time has passed.
4	<b>Assignments &amp; Projects:</b> Assignments and projects must be submitted on time. Students who delay their assignments or projects shall lose the mark allocated for the same.
5	<b>Cheating:</b> Cheating is an act of fraud that results in the cancelation of the student's exam or assignment. If it takes place in a final exam, the penalties stipulated for in the Uniform Students' Bylaw (2007) shall apply.
6	<b>Forgery and Impersonation:</b> Forgery/Impersonation is an act of fraud that results in the cancelation of the student's exam, assignment or project. If it takes place in a final exam, the penalties stipulated for in the Uniform Students' Bylaw (2007) shall apply.
7	<b>Other policies:</b> The University official regulations in force will be strictly observed and students shall comply with all rules and regulations of the examination set by the Department, Faculty and University Administration.



**Faculty of Medicine & Health Sciences.**  
**Department of Pharmacy**

**Bachelor of Pharm D**

**Course Plan (Syllabus) of Physical pharmacy**

**Course No. ( PHYP 105 )**

I. Information about Faculty Member Responsible for the Course:							
Name of Faculty Member:	Dr/ Sami Ahmed Al Sammit	Office Hours					
Location & Telephone No.:	773817898						
E-mail:	<a href="mailto:sami.alsammit@gmail.com">sami.alsammit@gmail.com</a>	SAT	SUN	MON	TUE	WED	THU



<b>II. Course Identification and General Information:</b>					
1	Course Title:	Physical Pharmacy			
2	Course Code & Number:	PHYP 105			
3	Credit Hours:	Credit Hours	Theory Hours		Lab. Hours
			Lecture	Exercise	
		2	2	--	--
4	Study Level/ Semester at which this Course is offered:	1 <sup>st</sup> Level / 2 <sup>nd</sup> Semester			
5	Pre –Requisite (if any):	-			
6	Co –Requisite (if any):	None			
7	Program (s) in which the Course is Offered:	Bachelor of Pharm D			
8	Language of Teaching the Course:	English			
9	Study System:	Semester based System			
10	Mode of Delivery:	Full Time			
11	Location of Teaching the Course:	Faculty of Medicine & Health Sciences			
12	Prepared by:	Dr/ Sami Ahmed Al Sammit			
13	Date of Approval:				

<b>III. Course Description:</b>
The course aims to provide students with basic principles of the physicochemical properties of pharmaceutical and various substances that are used in designing and preparation of different pharmaceutical dosage forms.

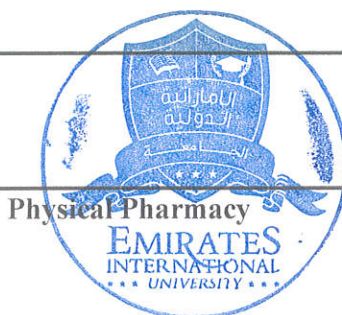
<b>IV. Course Intended Learning Outcomes (CILOs) :</b>
Upon successful completion of the Course, student will be able to:



	<b>A. Knowledge and Understanding:</b>
a1	Illustrate the physicochemical properties of pharmaceutical and various substances which are used in preparation of different pharmaceutical dosage forms.
a2	Outline properties of different pharmaceutical dosage forms and drug delivery systems.
a3	Discuss the principles of adsorption/desorption, interfacial tensions, and emulsifying, suspending and surface active agents.
a4	Identify principles and types of rheology.
a5	Explain the reaction kinetics and drug stability
	<b>B. Intellectual Skills:</b>
b1-	Relate the effect of physicochemical properties on formulation of pharmaceutical dosage forms.
b2-	Predict the order of reaction kinetics and drug degradation pathways
b3-	Explore the basics fundamentals of physical pharmacy in designing of different pharmaceutical dosage forms.
	<b>C. Professional and Practical Skills:</b>
c1-	Apply the rheological properties of some pharmaceutical substance to develop pharmaceutical preparations
c2-	Practice the solubility, partition coefficient and surface tension of some pharmaceutical substance.
c3-	Test the proper storage conditions based on drug degradation pathway
c4-	Calculate the reaction kinetic order and shelf life of some pharmaceutical substances
	<b>D. Transferable Skills:</b>
d1-	Implement writing and presentation skills and demonstrate creativity and time management.
d2	Work independently or collaboratively as a teamwork member to prepare seminars/ presentations or write reports

## V. Course Contents:

### A. Theoretical Aspect:



No.	Units/Topics List	Sub Topics List	Number of Weeks	Contact Hours
1	Introduction to physical pharmacy	Definition, orientation, process	1	2
2	State of matter, Solids	Crystal structure and external appearance, polymorphism, crystal hydrates, wetting of solid surfaces and powders dissolution of drugs Solid dispersions	1	2
3	Solubility and solution	Solvents for pharmaceutical aerosols, pH of drug solutions, Buffers	1	2
4	Factors influencing solubility	isotonic solutions, Diffusion of drugs in solution	1	2
5	Drug stability	Definition, factors stability of liquid and solids dosage forms	1	2
6	Reaction Kinetics and drug stability	Kinetics of chemical decomposition in solution Stability testing calculation of shelf-life	1	2
7	Surface and interfacial tensions	Definition , factors affecting Surface and interfacial tensions	1	2
8	Mid-term exam		1	2
9	Surface active agents	Definitions Some typical surfactants, applications	1	2
10	Emulsions, suspensions and other dispersed systems	Foams and defoamers	1	2
11	Polymers, drug absorption	Properties, Solution properties of polymers, Routes of administration	1	2
12	Physicochemical drug interactions and incompatibilities Complexes; classification and use.	Solubility problems pH effects in vitro and in vivo, Analysis of complexes.	1	2
13	Peptides, proteins and other biopharmaceuticals	Structure and solution properties of peptides and proteins, The stability of proteins and peptides	1	2
14	Adsorption at solid and liquid	Definition, types factors affecting	1	2



No.	Units/Topics List	Sub Topics List	Number of Weeks	Contact Hours
	interface.	Adsorption of drugs		
15	Rheology	Definition , classification and Application of polymers in drug delivery, Rheological characteristics of products	1	2
16	<b>Final Exam</b>		1	2
<b>Number of Weeks /and Units Per Semester</b>			<b>16</b>	<b>32</b>

### VI. Teaching Strategies of the Course:

Lectures , Discussions, Small group discussions, brainstorming, Group learning and Problem-based learning.

### VII. Assessment Methods of the Course:

Quizzes, Homework , report, short answers and Written exam

### VIII. Assignments:

No.	Assignments	Week Due	Mark	Aligned CILOs (symbols)
1	Assignment 1: Homeworks	9 <sup>th</sup>	5	a2,a3, a5, b2,b3, c3, d2
2	Assignment 1: Reports	13 <sup>th</sup>	5	a1-4, b1 c3,c4,d1, d2
<b>Total</b>			<b>10</b>	

### IX. Schedule of Assessment Tasks for Students During the Semester:

No.	Assessment Method	Week Due	Mark	Proportion of Final Assessment	Aligned Course Learning Outcomes
1	Assignments	9 <sup>th</sup> 13 <sup>th</sup>	5	10%	a1-5, b1-3, c3,c4, d1,d2

No.	Assessment Method	Week Due	Mark	Proportion of Final Assessment	Aligned Course Learning Outcomes
2	Quizzes 1 & 2	2 <sup>th</sup> and 11 <sup>th</sup>	5	10%	a2, a3, a5, b1, b3, c2
3	Mid-Term Theoretical Exam	8 <sup>th</sup>	20	20%	a1, a1, a3, a5, b1-3, c2-4
4	Final Theoretical Exam	15 <sup>th</sup>	70	60%	a1-5, b1-3, c1-4
<b>Total</b>			<b>100</b>	<b>100%</b>	

## X. Learning Resources:

- *Written in the following order:* Author, Year of publication, Title, Edition, Place of publication, Publisher.

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6. Martin, A., 2006, 'Physical Pharmacy – physical chemical principles in pharmaceutical science' 5th edition, Lippincott Williams & Wilkins., Philadelphia.

### 2- Recommended Readings and Reference Materials

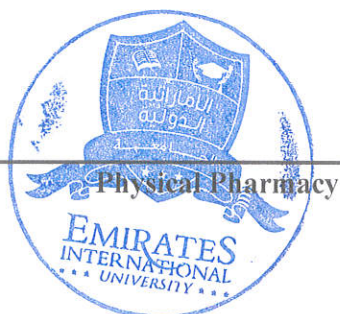
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8. Aulton, M.E, (2013) Pharmaceutics, the design and manufacture of medicines. 4<sup>th</sup> edition, Churchill Livingstone, Edinburgh.

### 3- Electronic Materials and Web Sites etc.:

[www.pubmed.com](http://www.pubmed.com)

<http://www.sciencedirect.com>

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### XI. Course Policies: (Based on the Uniform Students' By law (2007))

1	<b>Class Attendance:</b> Class Attendance is mandatory. A student is considered absent and shall be banned from taking the final exam if his/her absence exceeds 25% of total classes.
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